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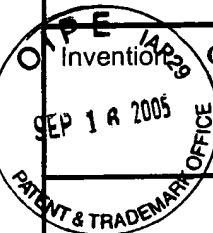
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# Chemical Amplification in the Design of Dry Developing Resist Materials

HIROSHI ITO and C. GRANT WILLSON

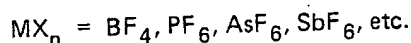
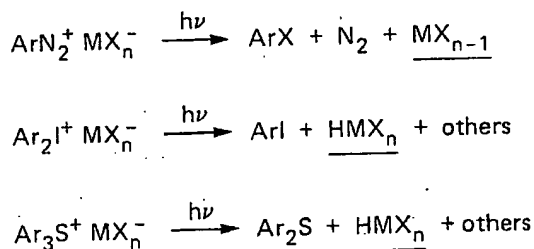
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A new resist system is described which undergoes spontaneous relief image formation. The resist is formulated from end capped poly(phthalaldehyde), PPA, and a cationic photoinitiator such as a diaryliodonium or triarylsulfonium metal halide. The extreme sensitivity of the resist is the result of designing for chemical amplification. The desired amplification results from the fact that photolysis of the sensitizer generates acid which catalyzes main chain cleavage of the polyaldehyde. The uncapped polymer is thermodynamically unstable with respect to reversion to monomer at room temperature so a single acid catalyzed scission results in complete depolymerization to volatile monomer. A single radiochemical event is thereby amplified in the sense that it produces an enormous number of subsequent chemical transformations. PPA/onium salt resist films are so sensitive that exposure to low doses of *e*-beam, X-ray or ultraviolet radiation results in complete self development without post-exposure processing of any kind. The exposed area simply vaporizes.

## INTRODUCTION

The function of the classical diazoquinone/novolac resists is based on a radiation induced structural transformation of the diazoquinone sensitizer which produces a large change in its efficiency as a dissolution inhibitor. The efficiency of the crucial photochemical transformation is characterized by the quantum yield for the process expressed as molecules transformed per photons absorbed. The quantum yield of typical diazoquinones is 0.2 to 0.3. Thus, three or four photons are required to transform a single molecule of sensitizer. This places a fundamental limit on the photosensitivity of such systems. In order to circumvent this intrinsic sensitivity limitation we have sought to design resist material systems that incorporate chemical amplification of the sort that characterizes the silver halide photographic emulsion systems. In these systems a single photo event initiates a cascade of subsequent chemical reactions that ultimately express the intended function.

Amplification of this sort is observed in several negative resist systems. One example of such a system involves the use of cationic photoinitiators in epoxy resins (1-6). Most of this work is based on the use of aryldiazonium salts, diaryliodonium salts or triarylsulfonium salts which generate strong acids upon exposure (Scheme 1). These acids initiate the cationic, ring opening polymerization of the oxirane residues that ultimately results in crosslinking. Here amplification is achieved by the catalytic nature of the initiation process. The acid is not consumed in the initiation step and can function to initiate several chains. Very sensitive nega-



Scheme 1. Photochemistry of onium salts.

tive resists have been developed that function on this basis (7).

We have explored the sensitized epoxy design in our laboratory. This work is exemplified by successful imaging of thick (6 $\mu$ ) films of Epi-Rez SU-8 epoxy resin (Celanese) by sensitizing the resin with *p*-hexyloxybenzene-diazonium hexafluorophosphate. Figure 1 shows high resolution negative images generated in such a formulation by exposure to 313 nanometers (nm) light and subsequent development in a nonpolar solvent.

We have recently focused our attention on the design of positive tone resist systems that incorporate chemical amplification. One such new class of positive resist materials is based on photochemical generation of a substance that catalyzes the cleavage of pendant groups on the side chains of appropriately functionalized polymers. Here, the effective quantum efficiency of the

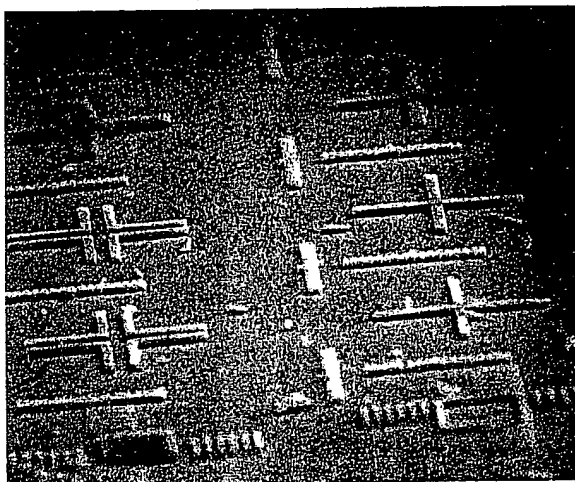


Fig. 1. Negative tone images printed in Celanese Epi-Rez SU-8 epoxy resin sensitized with *n*-hexyloxybenzenediazonium hexafluorophosphate exposed at 313 nm.

system is the product of the quantum yield of the photochemical event and the number of side chain transformations achieved by one molecule of the catalyst. This multiplicative response constitutes gain of the sort we seek.

A polymer upon which the new system is based is poly(*t*-BOC-*p*-hydroxystyrene), PBOCST. We have synthesized PBOCST both by chemical modification of commercially available poly(*p*-hydroxystyrene), Maruzen oil, and by radical polymerization of the appropriately substituted styrene monomer. The sensitizers for this system are again, various onium salts that undergo efficient photolysis to produce strong acids. Among the salts tested to date, triphenylsulfonium hexafluoroarsenate and diphenyliodonium hexafluoroarsenate are particularly effective.

The system functions, mechanistically, in the following way. Photolysis of the salts generates a local concentration of strong Brønsted acids. Upon post-exposure baking (100°) for a few seconds, the BOC group undergoes acidolysis to produce two volatile products, carbon dioxide and isobutylene, and liberate the phenolic hydroxyl. The resulting change in the polarity and solubility characteristics of the polymer is very large. Hence, the exposed areas can be selectively dissolved in polar solvents to provide positive tone images of the mask. Furthermore, if the developing solvent is nonpolar, the unexposed areas can be selectively dissolved to leave a negative tone image of the mask.

We have succeeded in producing excellent high resolution images in the new PBOCST resist (Fig. 2). It is very sensitive to deep UV radiation and can be dye sensitized for exposure in the mid UV or at conventional wavelengths. The materials are also sensitive to *e*-beam exposure (8).

We have now succeeded in designing a second stage of amplification into a positive resist material. In the conceptual design, one stage of amplification is achieved by choosing polymers which, when subjected to a single scission reaction, undergo spontaneous

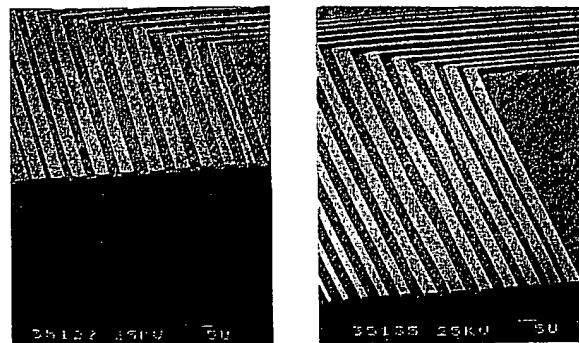


Fig. 2. PBOCST sensitized with diphenyliodonium hexafluoroarsenate and exposed at 254 nm. The image on the left is positive tone developed with MF312-water (1:1) and the image on the right is negative tone developed with CH<sub>2</sub>Cl<sub>2</sub>-hexane (2:1).

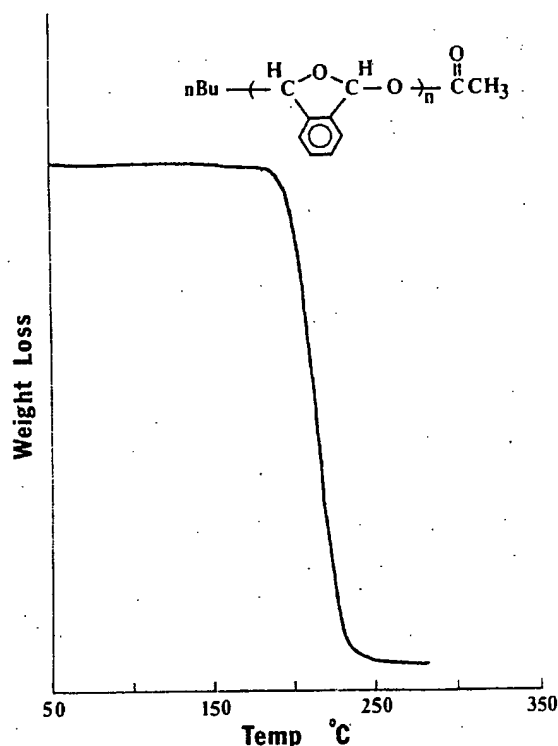
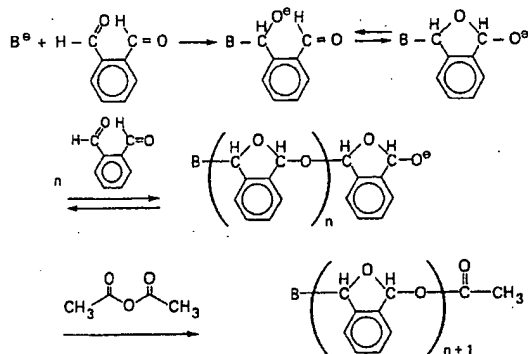


Fig. 3. Thermogravimetric analysis of polyphthalaldehyde synthesized by butyllithium initiation and end capped by acylation.

depolymerization that results in complete reversion to monomer. The second stage of amplification is achieved by including a substance which undergoes radiolysis to produce a product that catalyzes scission of the unstable polymer.

The polymer upon which the new system is based is a polyaldehyde. Aldehydes undergo anionic or cationic polymerization in a reversible, equilibrium reaction. Many of these systems have ceiling temperatures well below room temperature such that the polymerizations must be run at cryogenic temperatures and if the polymer is isolated, it rapidly depolymerizes to monomer. If, however, these polymers are end capped by acylation or alkylation prior to isolation and warming, they are of-

Dry developing resist materials have been sought for some time (15-17). The interest in such materials throughout the semiconductor industry stems from the potential they provide for higher throughput and higher yields. The higher yields result from both a reduction in process steps and the fact that such materials provide the potential for carrying our exposure, development, and pattern transfer under high vacuum, thereby reducing defects due to atmospheric contaminants and the development process. Dry processed X-ray negative resists that require plasma or RIE steps after exposure have been reported (16). Polymethacrylonitrile has been shown to dry develop upon heating after electron



Graph showing the effect of  $\gamma$ -ray dose on the methanol-insoluble fraction and molecular weight of polyisobutylene (No. 33). The x-axis represents  $\gamma$ -Ray Dose (Mrad) from 0 to 20. The left y-axis represents Methanol-Insoluble Fraction (%) from 0 to 100. The right y-axis represents Molecular Weight  $\times 10^{-4}$  from 0 to 6. Three data series are plotted: MW (triangles), Mn (squares), and Insoluble Fraction (circles). All three series show a decrease as dose increases.

$\gamma$ -Ray Dose (Mrad)	MW (Molecular Weight $\times 10^{-4}$ )	Mn (Molecular Weight $\times 10^{-4}$ )	Insoluble Fraction (%)
0	5.5	3.5	100
5	3.2	2.2	65
12	2.4	1.5	51
20	1.8	1.1	32

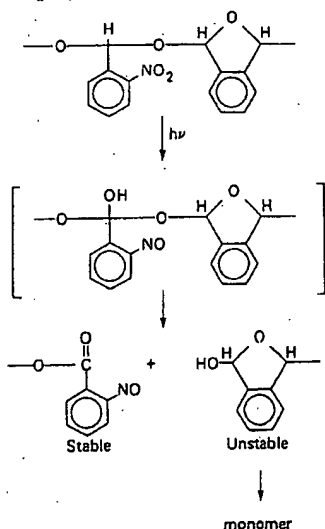
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unzipping occurs rapidly only when the scissioned polymer is dissolved.

Exposure of unsensitized PPA films to *e*-beam exposure followed by baking above 100° provides high resolution relief images at 5 to 20 microCoulomb ( $\mu\text{C}$ )/per square centimeter ( $\text{cm}^2$ ). However, the images are very shallow. Exposure to UV radiation results in approximately 15 percent loss in film thickness after heating. The UV imaging process is oxygen dependent. It does not occur when films are exposed *in vacuo*.

We reasoned that in order for the depolymerization to occur efficiently, a rational cleavage (scissioning) mechanism must be provided that generates a chain end that is a hemiacetal or hemiacetal ion in order to reestablish the polymerization equilibrium. Simple nonspecific cleavage to radical species should not be expected to produce the desired result. Consequently, we prepared copolymers of phthalaldehyde and *o*-nitrobenzaldehyde in attempt to increase the efficiency of the initial photo event. Exposure of copolymer films containing about 3 percent of *o*-nitrobenzaldehyde units to UV radiation resulted in relief image formation upon post-baking, with up to 60 percent loss in the film thickness in the exposed regions. We have found that photo-fragmentation did occur at the *o*-nitrobenzaldehyde acetal linkages as expected (18) (Scheme 3, Fig. 6). Fragmentation produced one hemiacetal chain end which depolymerized but the other photo fragment was shown to be terminated by an *o*-nitrosobenzoate linkage and therefore, efficiently capped (Scheme 3), as evidenced by the model reaction study (Fig. 7).

We next turned to sensitization of polyphthalaldehyde through incorporation of substances that produce strong Lewis or Brønsted acids upon radiolysis (Scheme 1) (1-6) in hope of achieving catalytic acidolysis of the acid-labile main chain acetal linkages (Scheme 4). Addition of certain cationic photoinitiators produced a dramatic result. For example, addition of triphenylsulfonium or diphenyliodonium metal halides (10 weight percent to the polymer) has allowed imaging of 1 $\mu$  thick



Scheme 3. Photosensitization of PPA by copolymerization with 2-nitrobenzaldehyde.

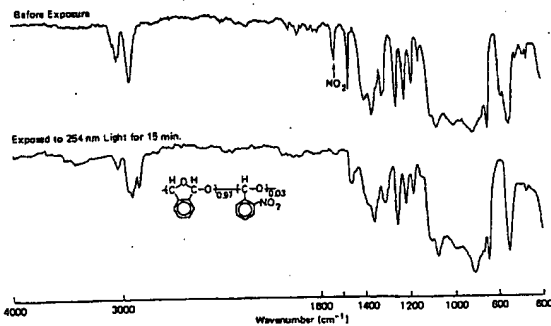


Fig. 6. Infrared spectra of poly(phthalaldehyde-co-2-nitrobenzaldehyde) before and after exposure to 254 nm light. Note the disappearance of the  $\text{NO}_2$  absorption.

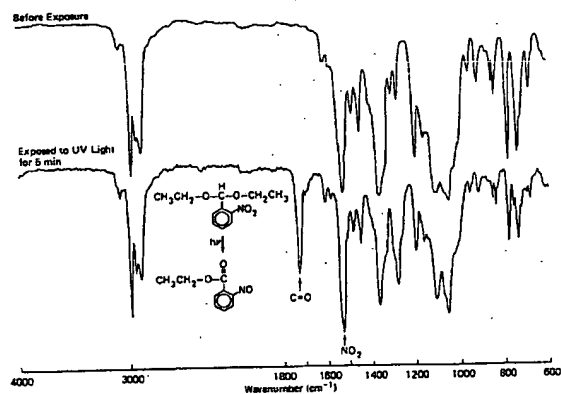
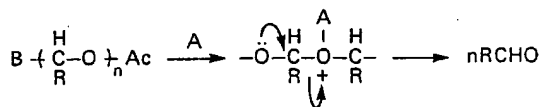
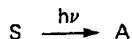


Fig. 7. Infrared spectra of the model compound 2-nitrobenzaldehyde diethylacetal before and after exposure to 254 nm light. Note the generation of the nitrosoester carbonyl absorption.



Scheme 4. Chemical amplification by sensitization with onium salts.

films of polyphthalaldehyde at 2 to 5 millijoule ( $\text{mJ}$ )/ $\text{cm}^2$  of deep UV radiation and at 1.0  $\mu\text{C}/\text{cm}^2$  of 20 keV electron beam radiation. Clean relief patterns with resolution below 1 $\mu$  line width are generated by exposure alone even without heating.

In a typical experiment, polyphthalaldehyde was dissolved in diglyme or cyclohexanone at 20 w/v% of solids, to which was added triphenylsulfonium or diphenyliodonium hexafluoroarsenate at 10 w/v% to the polymer. Films were spin-cast on silicon wafers, baked at 100°C for 10 minutes, and exposed. Optical micrographs of the resist images generated upon X-ray electron beam, and UV exposure are shown in Figs. 8, 9, and 10, respectively.

The UV sensitivity of the system depends on the loading and structure of the onium salt. For example, at 10 weight percent loading, the sensitivity with triphenylsulfonium hexafluoroarsenate (1.5 to 5  $\text{mJ}/\text{cm}^2$ ), diphenyliodonium hexafluoroarsenate (5  $\text{mJ}/\text{cm}^2$ ), and

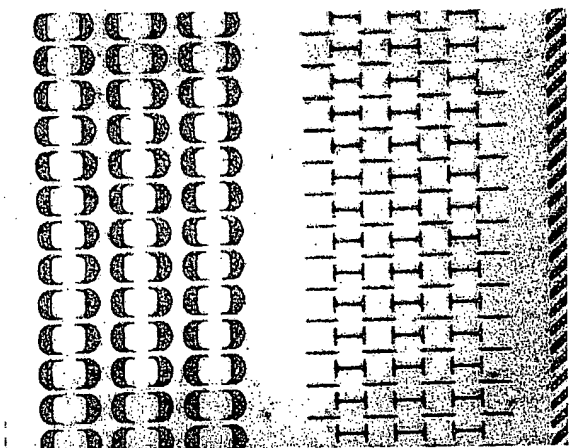


Fig. 8. Self developed PPA resist sensitized with diphenyliodonium hexafluoroarsenate exposed to Al-K $\alpha$  X-ray radiation. Minimum feature is 1.0 $\mu$ .

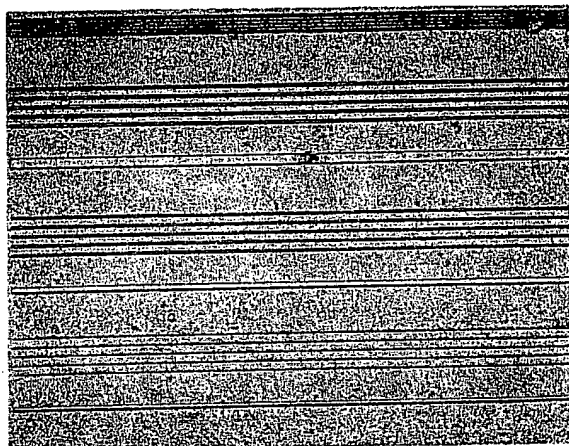
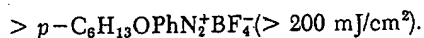
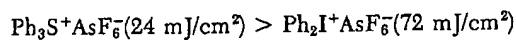


Fig. 9. Self developed PPA resist sensitized with diphenyliodonium hexafluoroarsenate exposed to 1  $\mu$ C/cm $^2$  of 20 keV electron beam radiation. Minimum written feature is 0.25 $\mu$ .

*p*-hexyloxybenzenediazonium tetrafluoroborate (6.5 mJ/cm $^2$ ) is similar, but at 2 weight percent loading a much higher dosage is required to achieve self development and the order of the sensitivity is as follows:



In any case, formation of monomeric phthalaldehyde due to acidolysis and depolymerization is clearly demonstrable by IR spectroscopy (Fig. 11).

Attempts to obtain scanning electron micrographs of PPA resist images has not been successful. The unexposed resist rapidly vaporizes in the microscope. We were surprised, therefore, to discover that electron beam generated relief images in PPA can be transferred into SiO $_2$  by either CF $_4$ /O $_2$  plasma etching (Figs. 12, and 13) or by treatment with buffered HF.

One might expect that resist systems as sensitive as PPA and, in particular, resist systems that incorporate catalytic species and chemical amplification might have very limited resolution. In theory, a single acid

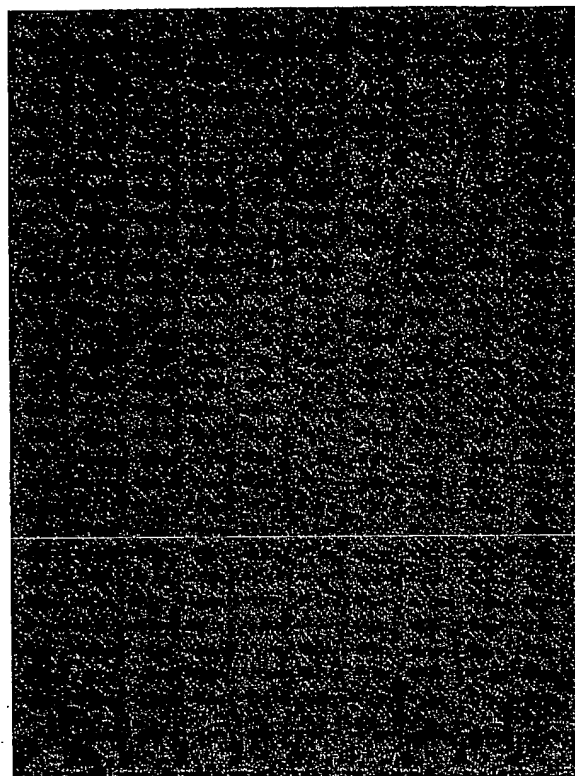


Fig. 10. Self developed PPA resist sensitized with triphenylsulfonium hexafluoroarsenate exposed to 2.4 mJ/cm $^2$  of deep UV (254 nm) radiation. Minimum feature is 0.7 $\mu$ .

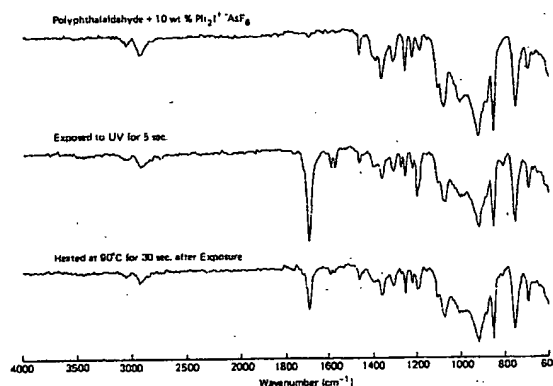


Fig. 11. Infrared spectra of PPA resist cast on sodium chloride before exposure, after very low dose exposure and after heating. Note generation of an intense aldehyde carbonyl due to monomer.

generating photo event should be sufficient to crosslink an entire film of epoxy resin or to depolymerize an entire PPA film. Fortunately, this is not observed experimentally as evidenced by the submicron images presented in the accompanying micrographs. Either the diffusion lengths of the acid catalytic species are very short in the glassy matrix or there exist a sufficient number of available terminating reactions to limit the lateral extent of the reactions in a film.

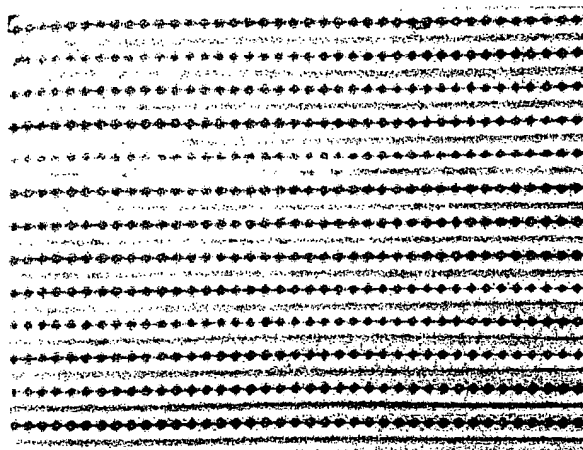


Fig. 12. Self developed patterns in  $1 \mu\text{C}/\text{cm}^2$  e-beam exposed PPA resist on  $\text{SiO}_2$ .

We are continuing to explore the limits of the PPA resist as well as other systems that incorporate chemical amplification.

### EXPERIMENTAL

Phthalaldehyde purchased from Aldrich Company was purified by recrystallization from  $\text{CH}_2\text{Cl}_2$ -hexane, and polymerized by anionic initiation with *n*-butyllithium, phenylmagnesium bromide, or potassium *t*-butoxide/18-crown-6 as initiator in tetrahydrofuran at  $-78^\circ\text{C}$  under high vacuum with use of a break-seal technique. All of the polymerizations were terminated by adding a cold mixture of acetic anhydride and pyridine at  $-78^\circ\text{C}$  to end-cap the polymer. As shown in Table 1, the end-capped polymers are of high molecular weight, stable to about  $180^\circ\text{C}$  according to TGA analysis, soluble in common organic solvents, and can be spin-cast from solution to provide clear, isotropic, noncrystalline films.

$^{60}\text{Co}$   $\gamma$ -radiation of the polymer powder was performed by The National Bureau of Standards (Washington, D. C.). The polymer powder was sealed in a glass tube under high vacuum and exposed at room temperature. Electron beam exposures were carried out with an IBM Vector Scan e-beam Exposure Tool at 20 keV. X-ray exposures were carried out under vacuum by  $\text{Al-K}_\alpha$  radiation, and UV exposures with Cannon PLA or Oriel illuminator in the contact mode.

IR spectra were recorded on a Perkin-Elmer 283 spectrometer and UV spectra on a Hewlett Packard 8450A UV/VIS spectrophotometer. Molecular weight determinations by GPC were made using a Waters Model 150 chromatograph equipped with six or eight

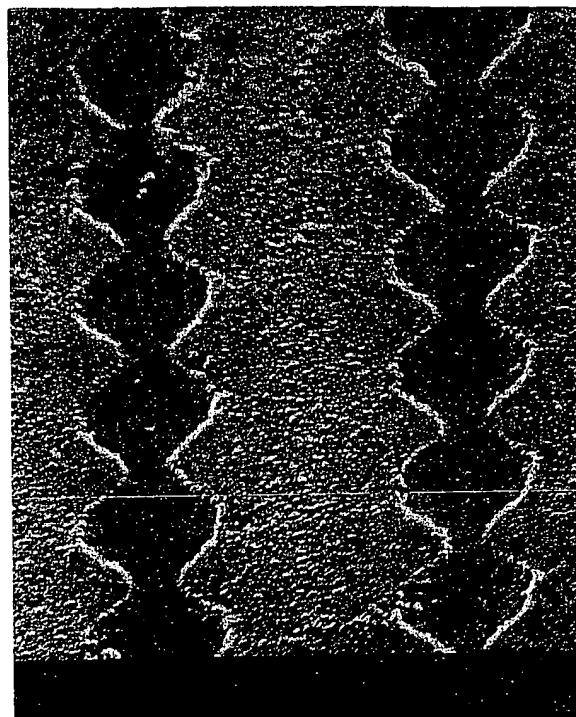


Fig. 13.  $\text{CF}_4$  plasma etched image transfer of resist pattern shown in Fig. 12.  $\text{SiO}_2$  thickness is 1000 Angstroms.

microstyrigel columns at 30 or  $40^\circ\text{C}$  in tetrahydrofuran. TGA measurements were performed on a DuPont 951 at a heating rate of  $10^\circ\text{C}/\text{min}$ .

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Table 1. Anionic Polymerization of Phthalaldehyde in THF at  $-78^\circ\text{C}$

Polymn. No.	Initiator (mole %)	Time hr	Yield %	$\text{Mn} \times 10^{-3}$	$\text{Mw} \times 10^{-3}$	D
30	PhMgBr (0.65)	22	89.5	54.3	97.1	1.79
33	nBuLi (0.33)	49	94.6	34.8	54.9	1.58
37	KOBu <sup>t</sup> /18-crown-6 (0.16)	72	78.9	30.8	98.5	3.20
64	nBuLi (1.18)	72	89.4	11.0	16.8	1.53
67	nBuLi (0.58)	72	90.5	36.9	43.0	1.17

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